

BLANK PAGE



IS: 10931 - 1984

Indian Standard SPECIFICATION FOR LAURIC ACID

UDC 661.7:547.295.72



@ Copyright 1984

INDIAN STANDARDS INSTITUTION.
MANAK BHAVAN, 9 BAHADUR SHAH ZAFAR MARG
NEW DELHI 110002

Indian Standard

SPECIFICATION FOR LAURIC ACID

Oils and Oilseeds Sectional Committee, CAFDC 5

Chairman

Representing

PROF M. M. CHAKRABARTY

Oil Technologists Association of India, Kanpur

Members

PROF A. C. GUPTA (Alternate to Prof M. M. Chakrabarty)

SHRIS. N. AGARWAL

Central Organization for Oil Industry and Trade, Bombay

SHRI H. P. GUPTA (Alternate) AGRICULTURAL MARKETING ADVISER TO THE GOVERNMENT OF INDIA

Directorate of Marketing and Inspection, Nagpur

SHRIR. I. VERMA (Alternate)

SHRI J. A. ASHTAPUTRE

Ministry of Defence (DGI)

SHRI P. N. AGARWAL (Alternate) DR B. P. BALIGA

Tata Oil Mills Co Ltd, Bombav

DR N. L. MURTY (Alternate) SHRI M. A. BHATT

Swastik Household and Industrial Products Ltd. Bombay

SHRI C. R. KRISHNAMURTHY (Alternate) DR N. V. BRINGI

Hindustan Lever Ltd, Bombay

DR V. V. S. MANI (Alternate) SHRI D. S. CHADHA

Central Committee for Food Standards (Ministry of Health and Family Welfare), New Delhi

SHRIMATI DEBI MUKHERJEE (Alternate)

SHRI P. K. CHARRABORTY DR K. C. GUHA

National Test House, Calcutta Central Food Laboratory, Calcutta

SHRI A. K. DHAR (Alternate) SHRI P. V. GUJARATHI

Khadi and Village Industries Commission, Bombay

SHRI V. LAKSHMIKANTHAN (Alternate : SHRI R. C. GUPTA

Directorate General of Technical Development. New Delhi

SHRIS. N. PANDEY (Alternate) SHRI S. P. GUPTA

Kusum Products Ltd, Rishra

SHRI A. R. N. KULU (Alternate) SHRI B. V. KANTAK

Godrei Soaps Ltd, Bombay

SHRI M. S. THAKUR (Alternate)

(Continued on page 2)

C Copyright 1984 INDIAN STANDARDS INSTITUTION

This publication is protected under the Indian Copyright Act (XIV of 1957) and reproduction in whole or in part by any means except with written permission of the publisher shall be deemed to be an infringement of copyright under the said Act.

IS: 10931 - 1984

(Continued from page 1)

Members

Representing

DR M. K. KUNDU

Directorate of Vanaspati, Vegetable Oils and Fats (Ministry of Civil Supplies), New Delhi

DR G. LAKSHM NARAYANA DR A. J. PANTULU (Alternate)

Regional Research Laboratory (CSIR), Hyderabad

DR B. M. LAL

Indian Agricultural Research Institute (ICAR), New Delhi

SHRI T. V. MATHEW

Central Agmark Laboratory, Nagpur

SHRI M. R. GROVER (Alternate)
SHRI R. K. MATHUR

Vanaspati Manufacturers' Association of India, Bombay

SHRI G. K. SOOD (Alternate) SHRI S. C. SINGHAL

Shriram Foods & Fertilizer Industries, New Delhi Shri T. N. AGGARWAL (Alternate)

SHRI G. V. SIRUR

Solvent Extractors Association of India, Bombay

SHRI L. KRISHAN KUMAR (Alternate)
DR (SHRIMATI) KAMLA SOHNIE CONSUI

Consumer Guidance Society of India, Bombay Regional Research Laboratory (CSIR),

Dr C. Srinivasulu F

Bhubaneshwar

DR (SHRIMATI) K. VIJAYALAKSHMI (Alternate)

SHRI S. D. THIRUMALA RAO Oil Technogical Research Institute, Anantapur SHRI G AZEEMODDIN (Alternate)

SHRI P. R. VISHWAMBHARAN Central Warehousing Corporation, New Delhi

SHRI P. R. VISHWAMBHARAN Central V SHRI K. GOPINATHA MENON (Alternate)

(Alternate)
Directorate of Oilseeds Development, Hyderabad

SHRI M. D. WASNIK
SHRI S. P. JUYAL (Alternate)
SHRI S. K. MATHUR,

Director General, ISI (Ex-officio Member)

Director (Chem)

Secretary

SHRI P. S. ARORA Sr Deputy Director (Chem), ISI

Oils and Fats Subcommittee, CAFDC 5: 1

Convener

Dr G. Lakshminarayana

Regional Research Laboratory (CSIR), Hyderabad

Members

AGRICULTURAL MARKETING Directorate of Marketing and Inspection, Nagpur Adviser to Government of

INDIA

SHRI T. V. MATHEW (Alternate)

SHRI J. A. ASHTAPUTRE Ministry of Defence (DGI)

. SHRI P. K. MAJUMDAR (Alternate)

ASSISTANT DIRECTOR (AGRI Indian Counil of Agricultural Research Institute, Engg) New Delhi

SHRI V. K. BANSAL

Central Organization for Oil Industry and Trade, Bombay

SHRI H. P. GUPTA (Alternate)

(Continued on page 11)

Indian Standard SPECIFICATION FOR LAURIC ACID

O. FOREWORD

- **0.1** This Indian Standard was adopted by the Indian Standards Institution on 30 March 1984, after the draft finalized by the Oils and Oilseeds Sectional Committee had been approved by the Chemical Division Council and the Agricultural and Food Products Division Council.
- 0.2 Lauric acid is a saturated fatty acid having 12 carbon atoms. Its molecular formula is CH₃ (CH₂)₁₀ COOH with 200.3 as its molecular weight. It is a crystalline solid with a faint characteristic fatty odour. It is a major fatty acid of coconut oil (Cocos nucifera Linn.), palm kernel oil (Elaeis guineensis Jacq.) and babassu oil (Orbignya martiana, O. oleifera or O. speciosa). Pisa fat (Actinodaphne angustifolia Nees or Actinodaphne hookeri Meissn) contains over 90 percent lauric acid. Commercially, lauric acid is obtained from lauric acid rich oils by fractional distillation of their fatty acids. By repeated fractionation steps, lauric acid of 99 percent purity can be made.
- **0.3** Lauric acid is a raw material for making lauryl alcohol, which is extensively used in the detergent industry. Its other uses include manufacture of non-yellowing alkyd resins, lauryl peroxide, ethanolamides, metallic salts, etc.
- **0.4** For the purpose of deciding whether a particular requirement of this standard is complied with, the final value, observed or calculated, expressing the result of a test or analysis, shall be rounded off in accordance with IS: 2-1960*. The number of significant places retained in the rounded off value should be the same as that of the specified value in the standard.

1. SCOPE

1.1 This standard prescribes the requirements and the methods of sampling and test for lauric acid.

^{*}Rules for rounding off numerical values (revised).

2. GRADES

2.1 The material shall be of three grades, namely, Grade 1, Grade 2 and Grade 3.

3. REQUIREMENTS

- **3.1 Description** Lauric acid shall be the product obtained by splitting lauric acid rich oils like coconut oil, palm kernel oil, etc, and fractionating the fatty acids suitably. It shall be clear when melted and free from adulterants, sediments, suspened and other foreign matter. Lauric acid may also contain minor proportion of C_8 - C_{18} fatty acids.
- 3.2 The material shall also comply with the requirement given in Table 1.

4. PACKING AND MARKING

- **4.1 Packing** The material shall be supplied in suitable containers, as agreed to between the purchaser and the supplier.
- **4.2 Marking** The containers shall be securely closed and legibly and indelibly marked with the following information:
 - a) Manufacturers' name and recognized trade-mark, if any;
 - b) Name and grade of the material;
 - c) Net mass of the material;
 - d) Batch number or lot number in code or otherwise; and
 - e) Month and year of manufacture.
- 4.2.1 The containers may also be marked with the ISI Certification Mark.

Note — The use of the ISI Certification Mark is governed by the provisions of the Indian Standards Institution (Certification Marks) Act and the Rules and Regulations made thereunder. The ISI Mark on products covered by an Indian Standard conveys the assurance that they have been produced to comply with the requirements of that standard under a well-defined system of inspection, testing and quality control which is devised and supervised by ISI and operated by the producer. ISI marked products are also continuously checked by ISI for conformity to that standard as a further safeguard. Details of conditions, under which a licence for the use of the ISI Certification Mark may be granted to manufacturers or processors, may be obtained from the Indian Standards Institution.

5. SAMPLING

5.1 Representative samples of the material shall be drawn and conformity of the material to the requirements of this specification shall be determined according to the procedure prescribed in Appendix B.

6. TESTS

6.1 Tests shall be carried out by the methods prescribed in col 6, 7 and 8 of Table 1.

TABLE 1 REQUIREMENTS FOR LAURIC ACID

(Clauses 3.2 and 6.1)

SL CHARACTERISTIC		REQUIREMENT FOR		METHODS	APPEN-	REF TO Indian	
No.		Grade 1	Grade 2	Grade 3	OF TEST, REF TO CL NO. IN IS: 548 (PART I)- 1964*	D1 x	INDIAN STANDARD
(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)
i)	Lauric acid, percent by mass, Min	99	90	70		_	IS: 548 (Part-3) 1976†
ii)	Moisture, percent by mass, Max	0.1	0.1	0.1	5.2		-
iii)	Saponification value	279-281	276-284	272-2 90	15		- .
iv)	Acid value shall not differ from saponification value by more than	4	4	4	7	. —	_
v)	Iodine value, Max	0.1	0.5	1.0	14	_	
vi)	Mineral acidity	Nil	Nil	Nil		A-1	
vii)	Ash, percent by mass, Max	0.01	0.01	0.02		A-2	
viii)	Unsaponifiable matter, percent by mass, Max	0.1	0.1	0.2	8		-
ix)	Titre	43-44	39-44	35 -4 3	12		_
x)	Colour, 5½ inch cell, $\Upsilon + 5 R$, Max	2	5	10	13		American .

^{*}Methods of sampling and test for oils and fats: Part 1 Sampling, physical and chemical tests (revised).

†Methods of sampling and test for oils and fats: Part 3 Analysis by gas liquid chromatography (being issued in parts).

APPENDIX A

(Table 1, Col 7)

TEST FOR MINERAL ACIDITY AND ASH

A-1. TEST FOR MINERAL ACIDITY

A-1.1 Quality of Reagents — Unless specified otherwise, pure chemicals and distilled water (see IS: 1070-1977*) shall be employed in tests.

Note — 'Pure chemicals' shall mean chemicals that do not contain impurities which affect the results of analysis.

A-1.2 Reagents

- **A-1.2.1** Methyl Orange Indicator 0.05 percent (m/v) solution.
- A-1.2.2 Light Petroleum Ether (60°C/80°C).
- A-1.3 Procedure Measure 10 ml of the melted sample into a separating funnel and shake intimately with three successive 10 ml portions of hot water. The temperature of the hot water should be more than the melting point of lauric acid. Combine the aqueous extracts, transfer to another separating funnel and remove traces of fatty acids in the water by extraction with light petroleum ether. Test the aqueous extract so obtained with a few drops of methyl orange indicator.
- **A-1.4** The material shall be taken to have satisfied the requirements of the test if the indicator does not show acid reaction.

A-2. DETERMINATION OF ASH

A-2.1 Apparatus

- A-2.1.1 Platinum Crucible
- **A-2.1.2** Desiccator containing an efficient desiccant, such as fused calcium chloride.
- A-2.2 Procedure Weigh accurately about 10 g of the air-dried material into a platinum crucible which has been previously dried, cooled in the desiccator and weighed. Heat the crucible over a low flame and ignite the contents gently. Incinerate the residue in a muffle furnace at 350°C ± 10°C until free from carbon. Cool the crucible in a desiccator and weigh. Repeat the above procedure of heating, cooling and weighing until the difference between two successive weighings does not exceed 1 mg.

^{*}Specification for water for general laboratory use (second revision).

A-2.3 Calculation

Ash, percent by mass =
$$\frac{100 \text{ m}}{M}$$

where

m =mass in g of the ash, and

M =mass in g of the material taken for the test.

APPENDIX B

(Clause 5.1)

SAMPLING OF LAURIC ACID

B-1. GENERAL REQUIREMENTS OF SAMPLING

- **B-1.0** In drawing, preparing, storing and handling test samples, the following precautions and directions shall be observed.
- B-1.1 Samples shall not be taken in an exposed place.
- B-1.2 The sampling instrument shall be clean and dry when used.
- **B-1.3** Precautions shall be taken to protect the samples, the material being sampled, the sampling instrument and the containers for samples from adventitious contamination.
- **B-1.4** To draw a representative sample, the contents of each container selected for sampling shall be mixed as thoroughly as possible by suitable means.
- B-1.5 The samples shall be placed in clean, dry glass stoppered bottles.
- **B-1.6** The sample containers shall be of such a size that they are almost completely filled by the sample.
- **B-1.7** Each sample container shall be sealed air-tight after filling and marked with full details of sampling, the date of sampling and the year and month of manufacture of the material.
- **B-1.8** Samples shall be stored in such a manner that the temperature of the material does not vary unduly from the normal temperature.

B-2. SCALE OF SAMPLING

- **B-2.0** Samples to determine conformity of the material to this specification shall be selected in accordance with the procedure given below. However, the purchaser and the supplier may agree to adopt any other procedure.
- **B-2.1 Lot** All the containers in a single consignment of one grade of the material drawn from a single batch of manufacture shall constitute the lot. If a consignment is declared to consist of different batches of manufacture, the batches shall be marked separately and the groups of containers in each batch shall constitute separate lots.
- **B-2.2** The number of containers to be selected from a lot shall depend upon the size of the lot and shall be in accordance with col 2 of Table 2.

TABLE 2 NUMBER OF CONTAINERS TO BE SELECTED FROM VARIOUS SIZES OF LOTS

LOT SIZE	Number of Containers to be Selected
N	n BE SELECTED
(1)	(2)
Up to 5	Alli (see Note)
6 to 65	5
66 to 110	7
111 and above	10

Note — When the lot size is 5 or less, the test results of each of the samples shall meet the corresponding requirement.

B-2.3 These containers shall be selected at random from the lot. In order to ensure the randomness of selection, procedure given in IS: 4905-1968* may be followed.

B-3. TEST SAMPLES AND REFEREE SAMPLE

- **B-3.1** Draw with an appropriate sampling instrument small portions of the material from different parts of the selected containers, the total quantity being sufficient to carry out the tests for all characteristics given in Table 1.
- **B-3.2** Mix thoroughly all portions of the material drawn from the same container to form an individual sample to represent the container. Equal quantities from the selected containers shall be mixed together to form a composite sample to represent the lot.

^{*}Methods for random sampling.

- **B-3.3** All the individual samples representing the selected containers and the composite sample representing the 10t shall be divided into three equal parts, thus forming three sets of test samples. These parts shall be immediately transferred to thoroughly dried bottles which shall then be sealed air-tight with glass stoppers. These shall be labelled with all the particulars of sampling given in **B-1.7**. One set of the test samples shall be sent to the purchaser and one to the supplier.
- **B-3.4 Referee Sample** The third set of the test samples, bearing the seals of the purchaser and the supplier, shall constitute the referee sample and shall be used in case of dispute between the purchaser and the supplier. It shall be kept at a place agreed to between the purchaser and the supplier.

B-4. NUMBER OF TESTS

- **B-4.1** The tests for saponification value, acid value and titre (see Table 1) shall be carried out on each individual sample of the set of test samples (see **B-3.3**).
- **B-4.2** The tests for the remaining characteristics given in Table 1 shall be carried out on the composite sample of the set of test samples (see **B-3.3**).

B-5. CRITERION FOR CONFORMITY

- **B-5.1** A lot shall be considered as conforming to this specification if it satisfies the criteria in **B-5.2.1** and **B-5.3** for the characteristics given in Table 1.
- **B-5.2** The test results for saponification value, acid value and titre shall be recorded as shown in Table 3. The mean and the range shall be calculated as follows and shall be recorded in col 4 and 5 respectively of Table 3:
 - Mean (\bar{X}) = $\frac{\text{The sum of the test results}}{\text{Number of test results}}$
 - Range (R) = The difference between the maximum and the minimum values of the test results.
- **B-5.2.1** The corrected mean as shown in col 6 of Table 3 shall be calculated. The lot shall be considered to have satisfied the requirement for a characteristic if the condition given in col 7 of Table 3 is satisfied.
- **B-5.3** The composite sample when tested for the remaining characteristics not tested in **B-5.2** shall satisfy the corresponding requirements for them as specified in Table 1.

IS: 10931 - 1984

TABLE 3 CRITERION FOR CONFORMITY (Clauses B-5.2 and B-5.2.1)						
SL No.	CHARACTERISTIC	TEST RE- SULTS 1, 2, 3	MEAN	RANGE	CORRECTED MEAN	CRITERION FOR CONFORMITY
(1)	(2)	(3)	(4)	(5)	(6)	(7)
i)	Saponification value		\vec{x}_1	R ₁	₹1-0.6 R1	Corrected mean specified value in Table 1 (iii)
ii)	Acid value		\overline{x}_2	R_2	\bar{x}_2 -0.6 R ₂	Corrected mean specified value in Table 1 (iv)
iii)	Titre		\overline{x}_3	R ₃	₹₃-0.6 R₃	Corrected mean specified value in Table 1 (ix)

(Continued from page 2)

Members

Representing

Dr N. V. Bringi

DR V. V. S. MANI (Alternate)

Shri D. S. Chadha

Central Committee for Food Standards (Ministry of Health and Family Welfare), New Delhi

SHRIMATI DEBI MUKHERJEE (Alternate)

SHRIR. D. KAWATRA

Directorate General of Technical Development, New Delhi

SHRI V. LAKSHMIKANTHAN

SHRI V. K. B. NAIR (Alternate) SHRI R. K. MARPHATIA

SHRI R. K. MATHUR

SHRI G. K. SOOD (Alternate)

DR N. L. MURTY

DR A. D. SHITOLE (Alternate)

DR S. M. PATEL

PROF V. V. R. SUBRAHMANYAM (Alternate)

DR J. V. PRABHAKAR .

SHRI R. D. SHENOI

DR I. S. SHENOLIKAR

DR RAMESH BHATT (Alternate) DR I. A. SIDDIQUI

DR R. A. KHAN (Alternate) SHRI M. S. THAKUR

SHRI S. D. THIRUMALA RAO SHRI D. ATCHYUTA RAMAYYA

SHRI M. D. WASNIK SHRI S. P. JUYAL (Alternate) Khadi and Village Industries Commission, Bombay

Indian Paint Association, Calcutta

Hindustan Lever Ltd, Bombay

SHRI RABIN SARKAR (Alternate)

D W MARHUR Vanaspati Manufacturers' Association of India, Bombay

Tata Oil Mills Co Ltd, Bombay

Oil Technologists' Association of India, Kanpur

Central Food Technological Research Institute (CSIR), Mysore Indian Confectionery Manufacturers' Association,

New Delhi Indian Council of Medical Research, New Delhi

Directorate of Vanaspati, Vegetable Oils and Fats (Ministry of Civil Supplies), New Delhi

Indian Soap and Toiletries Makers' Association, Bombay; and Godrej Soaps Ltd, Bombay Oil Technological Research Institute, Anantapur

(Alternate) Directorate of Oilseeds Development, Hyderabad

INTERNATIONAL SYSTEM OF UNITS (SI UNITS)

Rose	Units
DESC	United

QUANTITY	Unit	Symbol
Length	metre	m
Mass	kilo gr am	kg
Time	second	S
Electric current	ampere	Α
Thermodynamic temperature	kelvin	K
Luminous intensity	candela	$\mathbf{c}\mathbf{d}$
Amount of substance	mole	mol

Supplementary Units

QUANTITY	Unit	Symbol.
Plane angle	radian	rad
Solid angle	steradian	sr

Derived Units

QUANTITY	$\mathbf{U}_{ exttt{NIT}}$	SYMBOL	DEFINITION
Force	newton	N	$1 N = 1 kg.m/s^2$
Energy	joule	J	1 J = 1 N.m
Power	watt	w	1 W = 1 J/s
Flux	weber	Wb	1 Wb = 1 V.s
Flux density	tesla	T	1 $T = 1 \text{ Wb/m}^2$
Frequency	hertz	$H_{\mathbf{z}}$	$1 \text{ Hz} = 1 \text{ c/s (s}^{-1})$
Electric conductance	siemen	S	1 S = 1 A/V
Electromotive force	volt	V	1 V = 1 W/A
Pressure, stress	p a sca l	Pa	$1 Pa = 1 N/m^2$